

A stacked pyrazolo[3,4-*d*]pyrimidine-based flexible molecule: the effect on stacking of a bulky isopropyl group in comparison with a methyl/ethyl group¹

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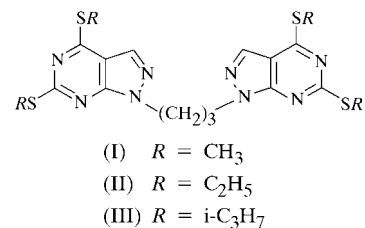
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In the crystal structure of 1,3-bis(4,6-diisopropylsulfanyl-1*H*-pyrazolo[3,4-*d*]pyrimidin-1-yl)propane, C₂₅H₃₆N₈S₄, the pairs of pyrazolo[3,4-*d*]pyrimidine rings in the molecule stack as a result of intramolecular π - π interactions between the heterocyclic rings. The crystal packing also exhibits an intermolecular C—H... π interaction between one methyl group of an isopropyl group and a pyrazolo[3,4-*d*]pyrimidine ring.

Comment

Interactions between aromatic moieties are known to play an important role in chemistry (Desiraju & Steiner, 1999; Hunter *et al.*, 2001; Tsuzuki *et al.*, 2002), stabilization of DNA/RNA structures (Hobza & Sponer, 1999), crystal engineering (Desiraju, 1995) and drug development (Meyer *et al.*, 2003). Use of a polymethylene and especially a trimethylene linker for demonstrating intramolecular stacking was pioneered by Browne *et al.* (1968) and early work was reviewed by Leonard (1979). In 1995, we reported the first synthesis (Avasthi *et al.*, 1995) and crystal structure determination (Biswas *et al.*, 1995) of a trimethylene-linker molecule, (I), based on the pyrazolo[3,4-*d*]pyrimidine core, which is isomeric with the biologically important purine system. The crystal structure exhibits both an unusual intramolecular stacking (U-motif) and intermolecular stacking. The robustness of the U-motif in (I) has been further demonstrated by the crystal structure determination of the ethyl analog, (II) (Avasthi, Rawat *et al.*, 2001), and other related propylene-linker compounds (Maulik *et al.*, 1998; Avasthi, Aswal & Maulik, 2001; Avasthi, Tewari *et al.*, 2002). Interestingly, no intramolecular stacking is observed when the trimethylene linker is replaced by an ethylene

(Avasthi, Rawat *et al.*, 2001) or tetramethylene linker (Maulik *et al.*, 2000; Avasthi, Farooq *et al.*, 2002). We report here the structure of 1,3-bis(4,6-diisopropylsulfanyl-1*H*-pyrazolo[3,4-*d*]pyrimidin-1-yl)propane, (III).



The conformation of (III) is shown in Fig. 1. The asymmetric unit contains only half of the molecule, which is related to the other half by the crystallographic symmetry operation $(1 - x, y, \frac{1}{2} - z)$. The molecule is folded at the centre of the bridge [C8—C9—C8A = 113.9 (2)°] as a result of intramolecular stacking between the pyrazolo[3,4-*d*]pyrimidine rings. For comparison, the corresponding angles in (I) and (II) are 114.1 (2) and 113.5 (2)°, respectively. In (III), as in (I) and (II), the two pyrazolo[3,4-*d*]pyrimidine rings are positioned in such a way that only part of the pyrimidine rings overlap (Fig. 1). The overlapping six-membered rings are separated by an average distance (the mean value of the distances of all the atoms in one ring from the least-squares plane through the atoms in the other ring) of 3.555 (2) Å [3.4 (4) and 3.37 (4) Å in (I) and (II), respectively], thus confirming the presence of an intramolecular π - π interaction. The pyrazolo[3,4-*d*]pyrimidine rings in (III), like those in (I) and (II), are nearly planar [maximum deviation = -0.036 (1) Å] and the angle between the least-squares planes is 21.96 (4)° [13.2 (1)° in both (I) and (II)]. It appears that because the isopropyl group is bulkier than the methyl/ethyl groups, the average intramolecular distance [3.555 (2) Å] and angle [21.96 (4)°] between the least-squares planes have increased appreciably. However, the most striking effect of the presence of the isopropyl group instead of the methyl/ethyl groups found in (I) and (II) is seen

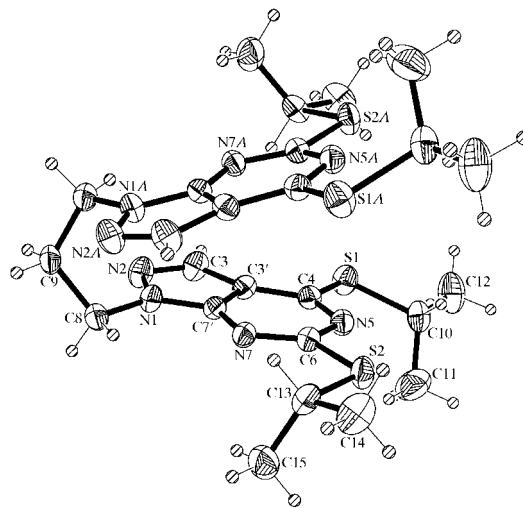


Figure 1
Displacement ellipsoid plot (30% probability) showing the molecular structure of (III) and the atomic labelling scheme. [Symmetry code: (A) $1 - x, y, -z + \frac{1}{2}$]

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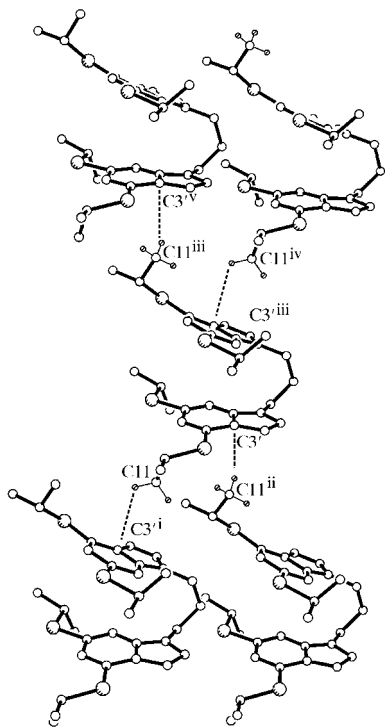


Figure 2
Crystal-packing diagram of (III), showing the intermolecular C—H... π -stacking interactions (dashed lines) between the pyrazolo[3,4-*d*]pyrimidine rings. [Symmetry codes: (i) $\frac{3}{2} - x, \frac{1}{2} - y, \frac{1}{2} - z$; (ii) $\frac{3}{2} - x, \frac{1}{2} + y, \frac{1}{2} - z$; (iii) $1 - x, y, \frac{1}{2} - z$; (iv) $x - \frac{1}{2}, \frac{1}{2} + y, z$; (v) $x - \frac{1}{2}, y - \frac{1}{2}, z$.]

in the packing diagram of (III) (Fig. 2). Molecule (III) does not exhibit intermolecular stacking due to the π - π interaction but instead exhibits an intermolecular C—H... π interaction (Fig. 2) between one methyl group of an isopropyl group and the adjacent C3' atom of the pyrazolo[3,4-*d*]pyrimidine ring (H...C3' = 3.14 Å). In conclusion, replacement of the methyl or ethyl group in (I) or (II) by a bulky isopropyl group does not seriously affect the robustness of the U-motif formed by intramolecular stacking due to the aromatic π - π interaction. However, the parallel mode of intermolecular stacking seen in (I) and (II) is not present in (III) but is replaced by a C—H... π interaction.

Experimental

Compound (III) was prepared by stirring a mixture of 4,6-diisopropylsulfanyl-1*H*-pyrazolo[3,4-*d*]pyrimidine and 1,3-dibromopropane in dimethylformamide in the presence of anhydrous potassium carbonate (Avasthi *et al.*, 1995). Diffraction-quality crystals were prepared by slow evaporation from a 1:1 solution of methanol and butanol at room temperature.

Crystal data

$C_{25}H_{36}N_8S_4$	$D_x = 1.235 \text{ Mg m}^{-3}$
$M_r = 576.90$	Mo $K\alpha$ radiation
Monoclinic, $C2/c$	Cell parameters from 44 reflections
$a = 21.044 (2) \text{ \AA}$	$\theta = 3.9\text{--}12.5^\circ$
$b = 8.814 (1) \text{ \AA}$	$\mu = 0.34 \text{ mm}^{-1}$
$c = 18.953 (2) \text{ \AA}$	$T = 293 (2) \text{ K}$
$\beta = 118.07 (1)^\circ$	Rectangle, colourless
$V = 3101.9 (6) \text{ \AA}^3$	$0.30 \times 0.28 \times 0.20 \text{ mm}$
$Z = 4$	

Data collection

Bruker P4 diffractometer	$h = -24 \rightarrow 1$
θ -2 θ scans	$k = -1 \rightarrow 10$
3403 measured reflections	$l = -20 \rightarrow 22$
2738 independent reflections	3 standard reflections
2352 reflections with $I > 2\sigma(I)$	every 97 reflections
$R_{\text{int}} = 0.020$	intensity decay: none
$\theta_{\text{max}} = 25.0^\circ$	

Refinement

Refinement on F^2	$(\Delta/\sigma)_{\text{max}} = 0.001$
$R[F^2 > 2\sigma(F^2)] = 0.035$	$\Delta\rho_{\text{max}} = 0.22 \text{ e \AA}^{-3}$
$wR(F^2) = 0.094$	$\Delta\rho_{\text{min}} = -0.20 \text{ e \AA}^{-3}$
$S = 1.03$	Extinction correction:
2738 reflections	<i>SHELXL97</i>
173 parameters	Extinction coefficient:
H-atom parameters constrained	0.0045 (3)
$w = 1/[\sigma^2(F_o^2) + (0.0402P)^2 + 1.8299P]$	
where $P = (F_o^2 + 2F_c^2)/3$	

Data collection: *XSCANS* (Siemens, 1996); cell refinement: *XSCANS*; data reduction: *XSCANS*; program(s) used to solve structure: *SHELXTL* (Bruker, 1997); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: DE1217). Services for accessing these data are described at the back of the journal.

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